

Pipe Corrosion in the Vicinity of a Major Soda Bottling Plant

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Identification of Problem

- HRSD and one of its member jurisdictions are experiencing significant problems with line corrosion in an area down-line from a major soda bottling plant.



Photo by: Eric Jackson, HRSD



Photo by: David Creque, HRSD

Line Corrosion

- The corrosion was atypical



Photo by: Eric Jackson, HRSD

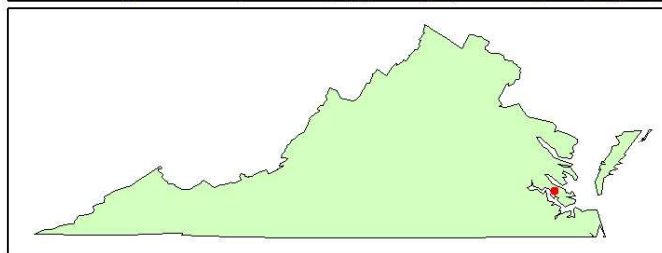
Purpose

- To determine cause of the underlying problems associated with the corrosion of sewer lines in the vicinity of a major soda bottling plant
- Work with bottling plant to implement pretreatment modifications/additions to prevent future corrosion

Study Area

- Southeast Virginia
- Study Area is mixed residential & commercial
 - Soda bottler discharges to city collection system→
City pump station→
City force main→
HRSD Interceptor System→
HRSD Treatment Plant

Map of Study Area



Sampling Activities

- Continuous pH Monitoring
 - MH ~1200 feet down-line of soda bottler
 - MH at influent to city pump station
 - ~3900 feet from bottler
 - Simultaneous at MH outside soda bottler and MH at influent to city pump station
- Alkalinity grab samples
 - Prompted by results from pH monitoring
 - MH outside soda bottler
 - MH ~1700 feet down-line of soda bottler
 - MH at influent to city pump station
 - HRSD Pump Station Wet Well (~5.75 miles from soda bottler)
 - HRSD Treatment Plant RWI (~8.5 miles from soda bottler)

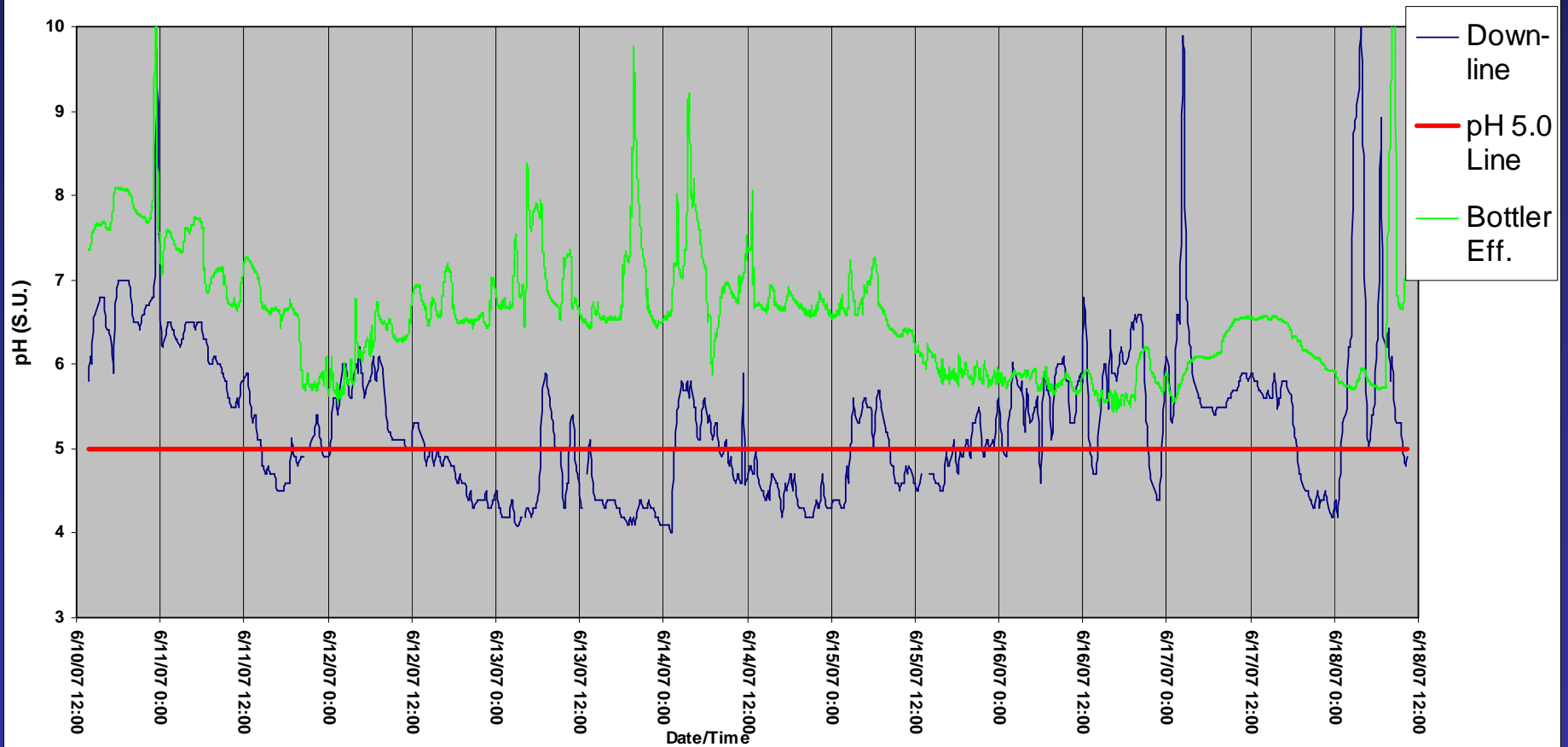
pH Results

~1200 feet Down-line of Bottler

- Deployed from 6/10/07 through 6/18/07
- n=751
 - Average = 5.3 SU
 - Minimum pH = 4.0 SU (6/14/07 / 0100-0115)
 - Maximum pH = 10.2 SU (6/18/07 / 0345)
- 45 % of the samples were less than 5.0 SU
- The bottler's effluent pH was consistently above 5.4 SU

pH Results

Down-line of Soda Bottler



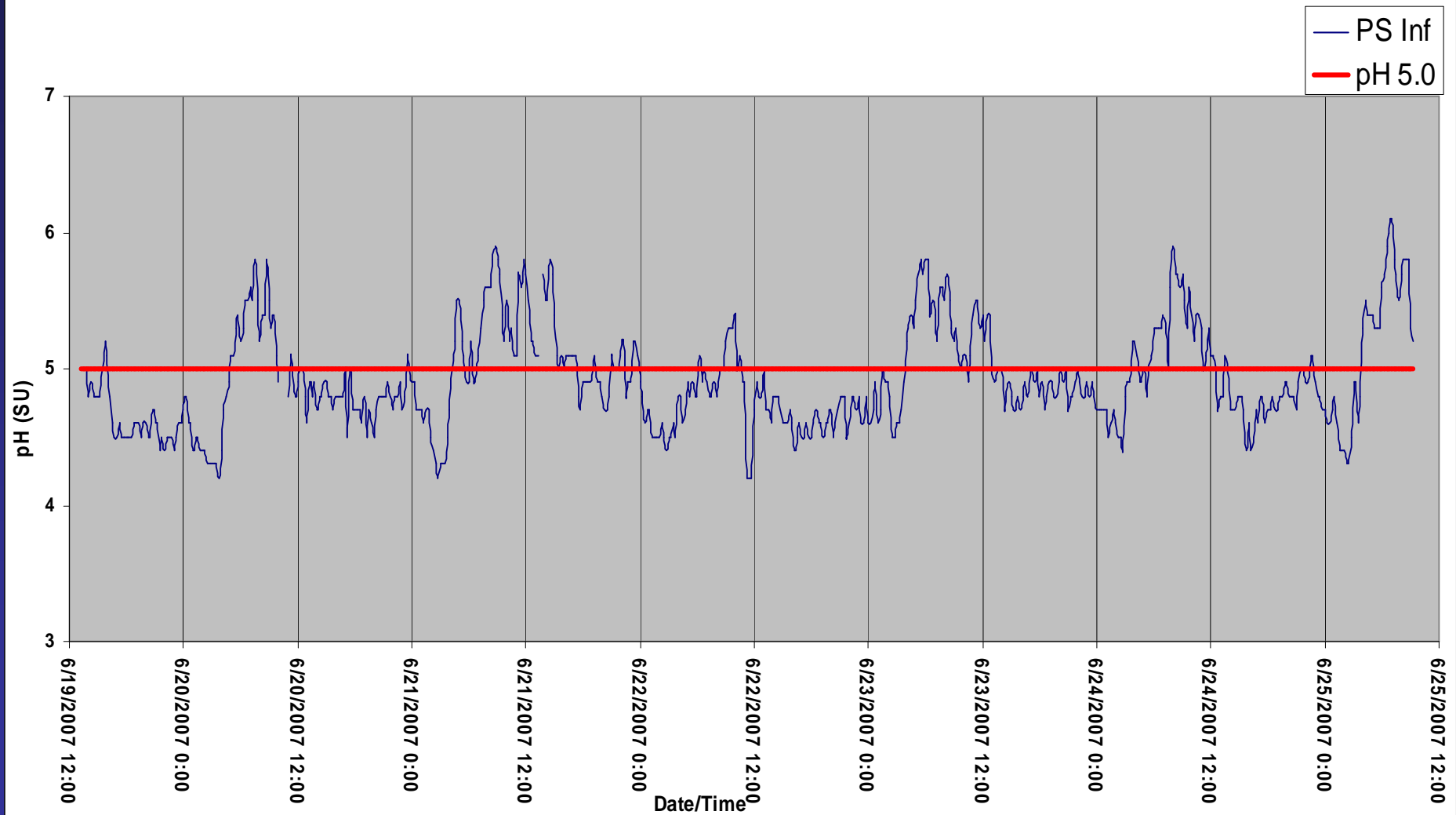
pH Results

City Pump Station Influent

- ~3900 feet from the bottler
- Deployed from 6/19/07 through 6/25/07
- n=557
 - Average = 4.9 SU
 - Minimum = 4.2 SU (6/22/07 / 1115-1145)
 - Maximum = 6.1 SU (6/25/07 / 0700)
- 70% of the samples were less than 5.0 SU

pH Results

City Pump Station Influent pH



pH Results

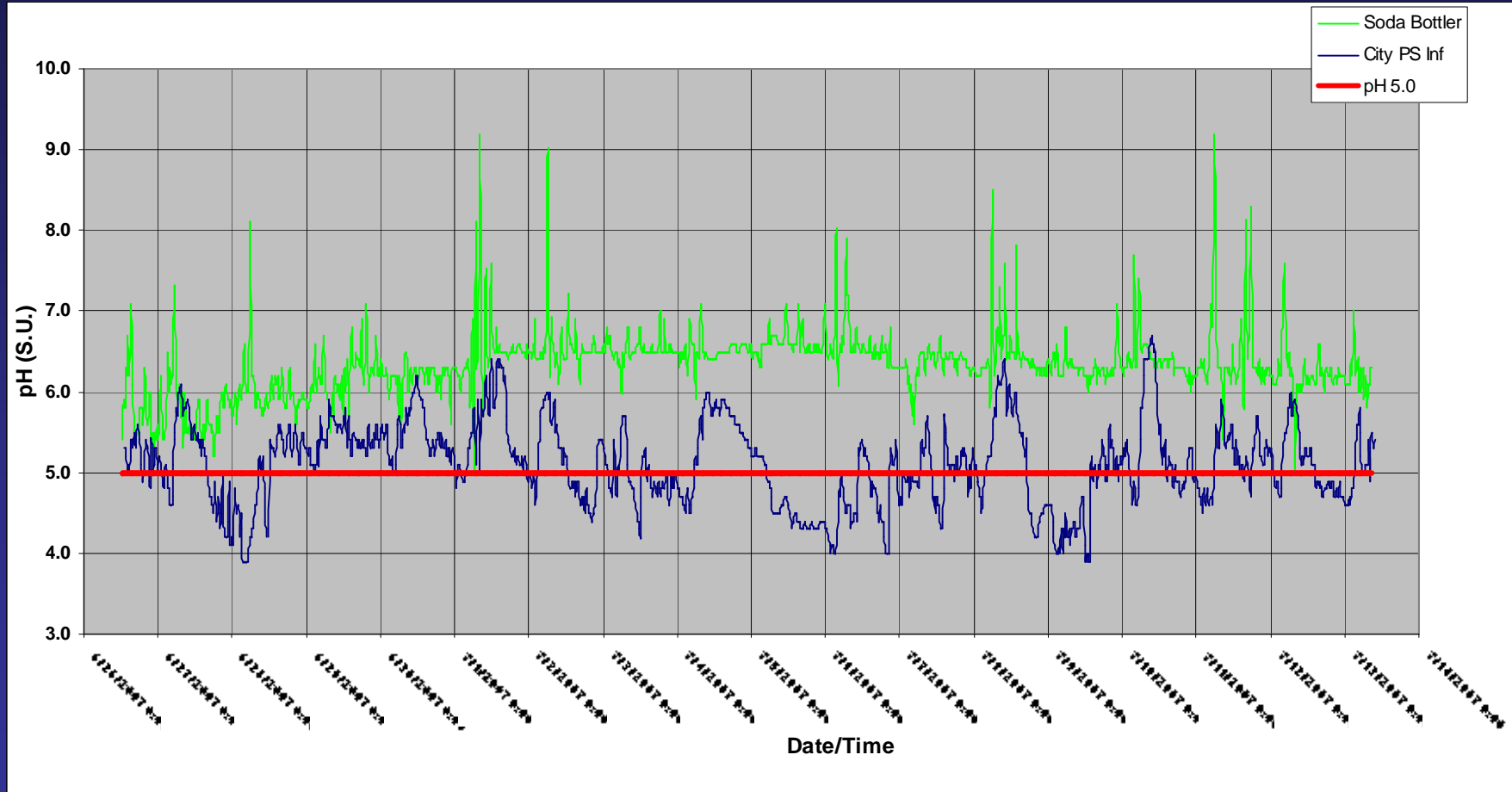
Simultaneous Comparison

- Compared discharge from soda bottler to city PS influent from 6/26/07 through 7/13/07
- n= 1619

	Bottler	City PS Inf.
Average pH	6.3	5.1
Minimum pH	5.0	3.9
Maximum pH	9.2	6.7
% of samples <5.0 SU	0%	38 %

pH Results

Simultaneous Comparison



Alkalinity Results

- Decreased between bottler and ~1700 feet down-line
- Started to increase after commingling with other wastestreams

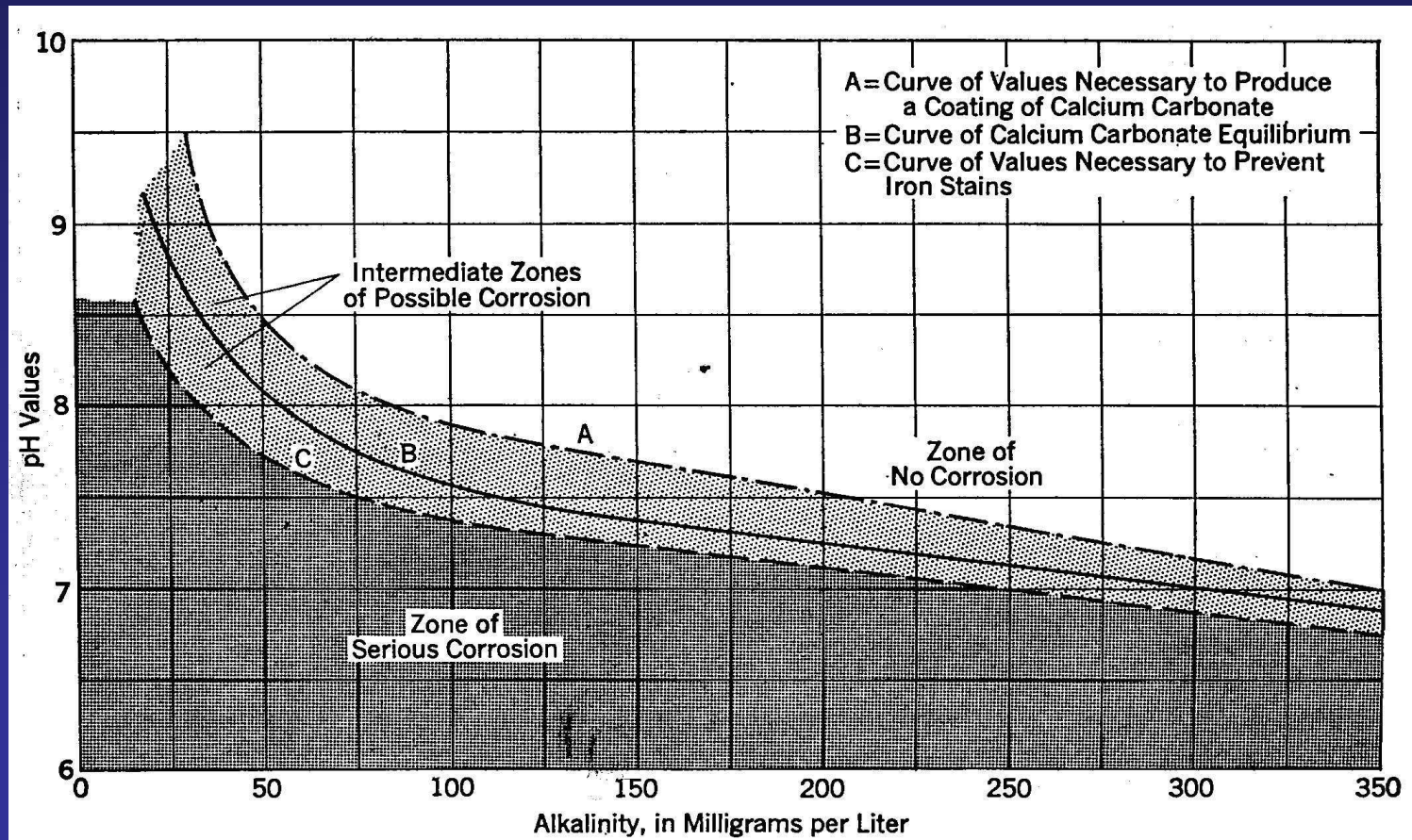
Date	Bottler	1700 ft	City PS	HRSD PS	TP RWI
6/25/07	121	107	91	NS	NS
7/9/07	38	39	81	NS	NS
8/29/07	50	18	48	144	220
9/12/07	70	33	55	182	200

All concentrations are in mg/L of CaCO_3

NS= No Sample taken

Conclusions

- Low pH and low alkalinity are the probable causes of the interior corrosion in the HRSD and city-owned force mains.
 - “The lower the alkalinity, in milligrams per liter, the higher the pH value must be to prevent corrosive action.” (Hardenbergh 1960)



Hardenbergh 1960

Conclusions

- The drop in pH and alkalinity can be attributed to:
 - Sugars from soda product (high BOD, citric acid) decompose producing organic acids. This in turn lowers the pH and could cause a further drop in alkalinity.
 - Retention time in soda bottler's equalization tank (5750 gallon tank, 10-20 minutes) is not sufficient to allow the solution to reach equilibrium following the addition of sodium hydroxide. The pH and alkalinity decrease as the solution approaches equilibrium further down the line.

Recommendations

- Study is on-going
 - Recently requested further information from the bottler
- Researching possible solutions
 - May include:
 - Increase pH permit limit
 - Increase alkalinity of wastestream prior to discharge
 - Increase retention time in equalization tank
 - Use existing processes (cleaning) to their advantage

Works Cited

- Hardenbergh, W.A., and Rodie, E.S. Water Supply and Waste Disposal. International Textbook Co, Scranton, PA 1960.
- HDR Engineering Inc. “Lee Hall Sanitary Force Main Failure Evaluation.” July 2004.

Acknowledgements

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Questions?

